

Simulation of Dendritic Solidification in Cubic and HCP Crystals by Cellular Automaton and Phase-Field Models

Mohsen Asle Zaeem*

(Ph.D., Assistant Research Professor)

Center for Advanced Vehicular Systems, Mississippi State University

Hebi Yin

(Ph.D., Postdoctoral Research Associate)

Oak Ridge National Laboratory

Sergio D. Felicelli

(Ph.D., Professor)

Mechanical Engineering Department, Mississippi State University

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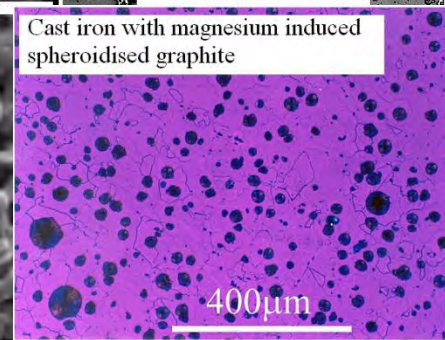
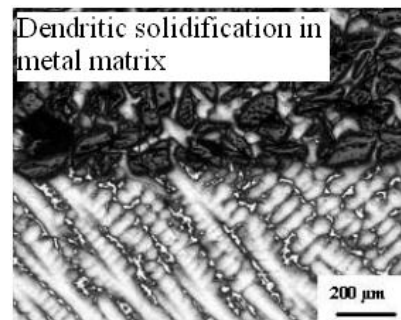
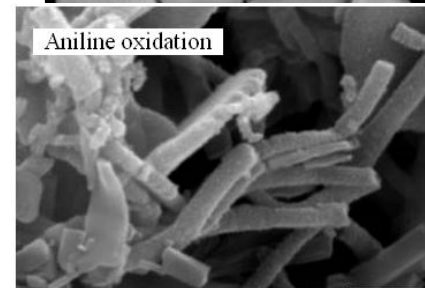
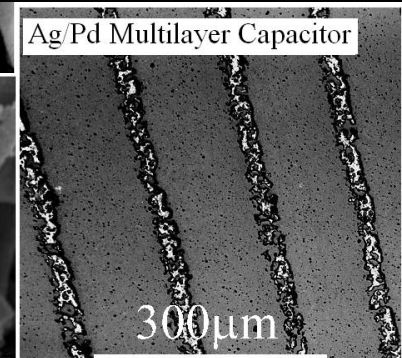
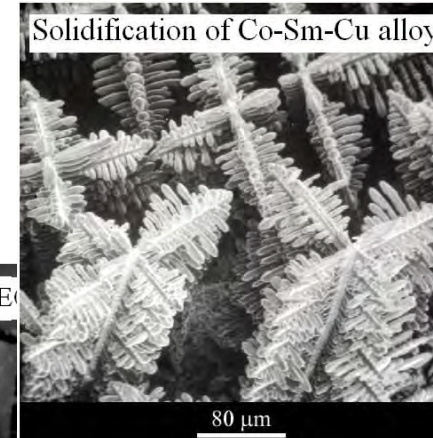
Microstructures are the shape and alignment of the microscopic components of materials and may consist of:

- Spatially distributed phases of different compositions and/or crystal structures
- Grains of different orientations, domains of different structural variants
- Domains of different electrical or magnetic polarizations
- Structural defects

Microstructural evolution is common in many fields including biology, hydrodynamics, chemistry, and phase transformations.

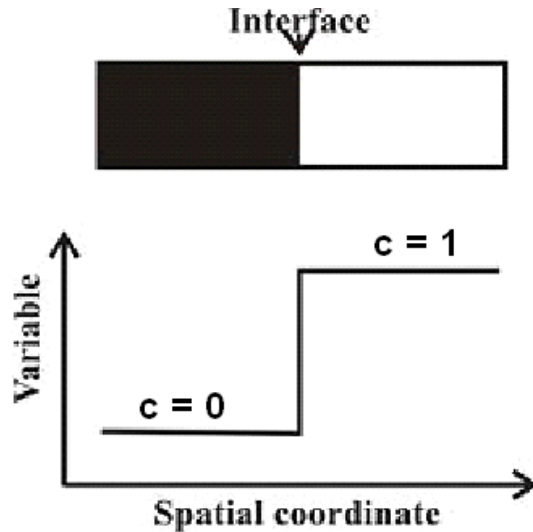
Microstructures controls

- Mechanical
 - Electrical
 - Magnetic
 - Optical
 - Chemical
- properties of materials

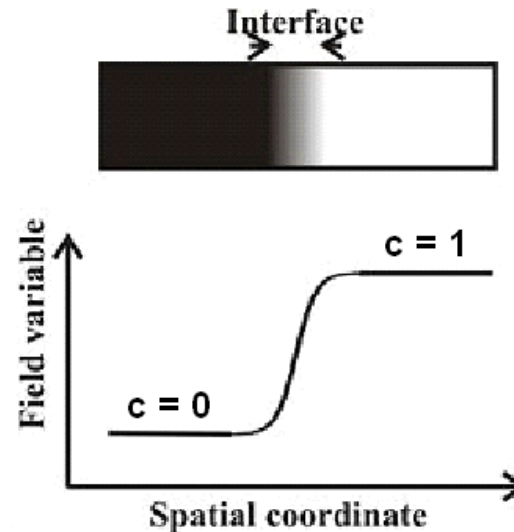


The study of free boundaries-interfacial regions can be grouped into two categories

Sharp-interface models



Phase-field models (diffusive interface)



Non-conserved phase-field

$$\dot{\phi} = D_2(\phi)$$

Solidification and melting

Conserved phase-field

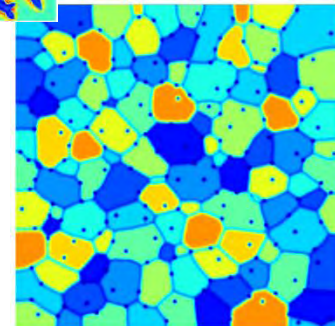
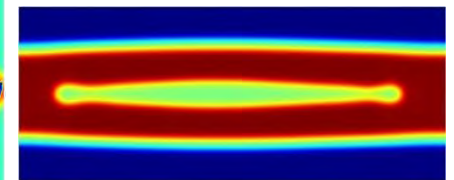
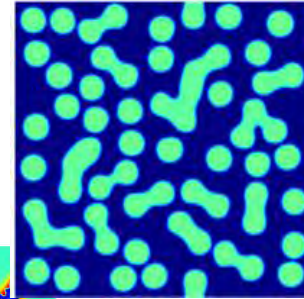
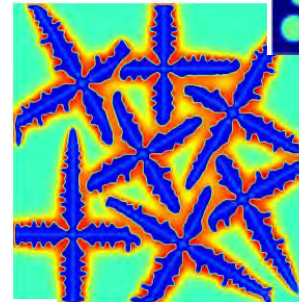
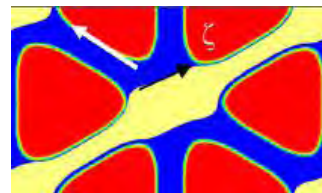
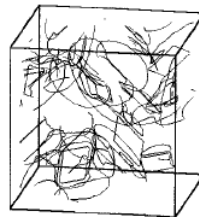
$$\dot{\phi} = D_4(\phi)$$

- Solid-state phase transformations (coupled to elasticity)
- Interfaces between immiscible fluids (coupled to Navier–Stokes)

L.-Q. Chen, *Annu. Rev. Mater. Res.* **32** (2002) 113.
G. Caginalp and W. Xie, *Phys. Rev. E* **48** (1993) 1897.

Applications of Phase-Field Model

- Solid State Phase Transformation
- Phase Transformation in Thin Films
- Solidification
- Grain Growth polycrystalline materials
- Dislocation Dynamics
- Crack Propagation
- Electromigration
- Multi-Phase Fluid Flow
-



Long-Qing Chen, Phase-field models of microstructure evolution, *Annu. Rev. Mater. Res.* 32 (2002) 113-140.

Yunzhi Wang, Ju Li, Overview No. 150: Phase field modeling of defects and deformation, *Acta Materialia* 58 (2010) 1212–1235.

Commonly Used Numerical Methods to Solve The Governing Equations of Phase-Field Models

- Finite difference method (FDM) (Cahn and Kobayashi 1995; Johnson 2000)
- Fourier-spectral methods (Chen and Shen 1998, 2009; Boisse et al. 2007)

These methods have limitations in 2D and 3D models when irregular geometries or complex boundary conditions are involved



We develop a general numerical method applicable to a variety of geometries and boundary conditions.

M. Asle Zaem and S.Dj. Mesarovic, Journal of Computational Physics 229 (2010) 9135-9149.

Dendritic Solidification of Cubic and Hexagonal Materials

During solidification of metals and their alloys, the formation of complex dendrite microstructures has significant effects on the mechanical and material properties of cast alloys.

Dendritic Solidification of Cubic and Hexagonal Materials

Cellular automaton

Heat transfer: $\frac{\partial T}{\partial t} = \alpha \cdot \nabla^2 T + \frac{L}{\rho C_p} \frac{\partial f_s}{\partial t}$

α → thermal diffusivity

Mass transfer: $\frac{\partial C_i}{\partial t} = D_i \cdot \nabla^2 C_i + C_i \cdot (1 - k) \frac{\partial f_s}{\partial t}$

Solid/Liquid interface: $C_s = k \cdot C_l$

k → partition coefficient

The increase solid fraction: $\Delta f_s = (C_1^* - C_1) / (C_1^* \cdot (1 - k))$

Interface equilibrium composition:

equilibrium liquidus T at C_0 Gibbs-Thomson coefficient

$C_1^* = C_0 + \frac{T^* - T_1^{eq} + \Gamma K f(\varphi, \theta_0)}{m_1}$

interface equilibrium T Γ → curvature at SL interface

m_1 → liquidus slope

Function of anisotropy of surface tension:

$f(\varphi, \theta_0) = 1 - \delta \cos[4(\varphi - \theta_0)]$

growth angle δ → anisotropy coefficient


→ preferential direction

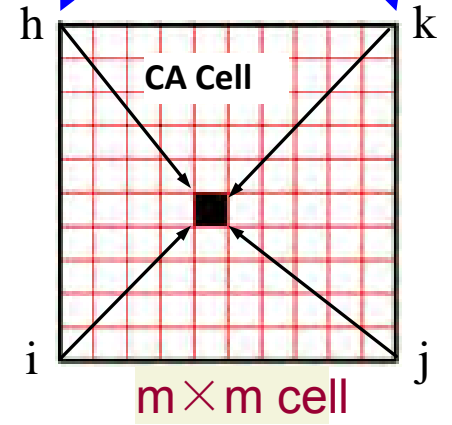
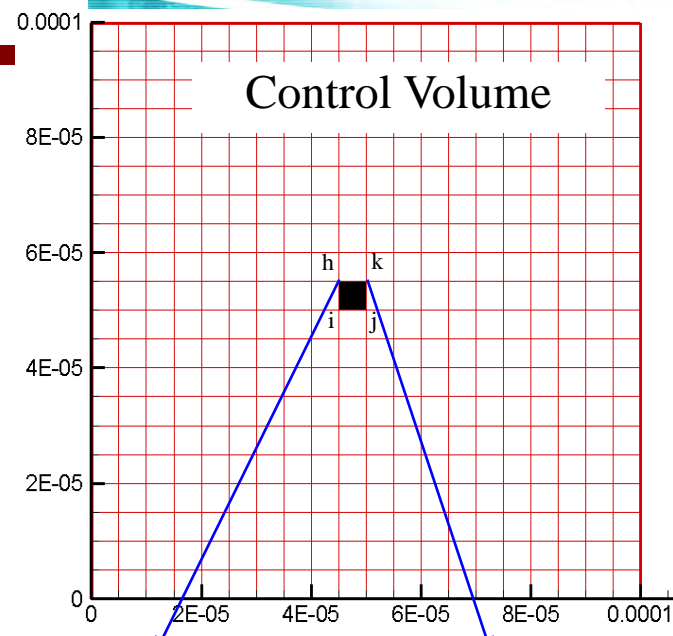
Time step for heat:



$\Delta t_T = \frac{\rho C_p (m \cdot \Delta x)^2}{4.5 \lambda}$

$m \cdot \Delta x$ → cell size






Time step for mass: $\Delta t_C = \frac{\Delta x^2}{4.5 D_1}$

Ration of two time steps:

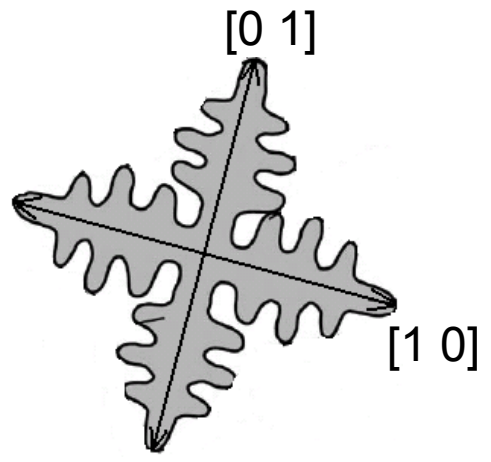
$N_t = \Delta t_C / \Delta t_T$



Dendritic Solidification of Cubic and Hexagonal Materials

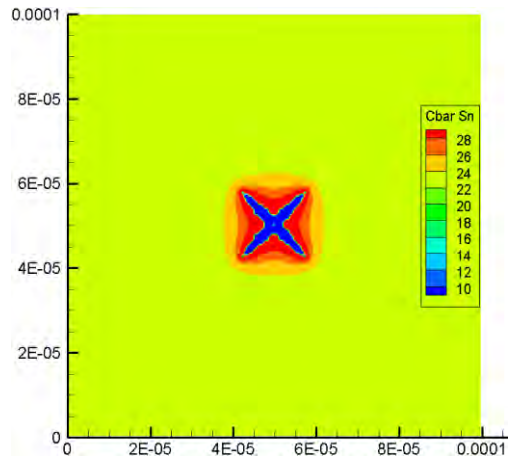
Cellular automaton (Pb-23%Sn)

Pb, cubic structure with four-fold symmetry

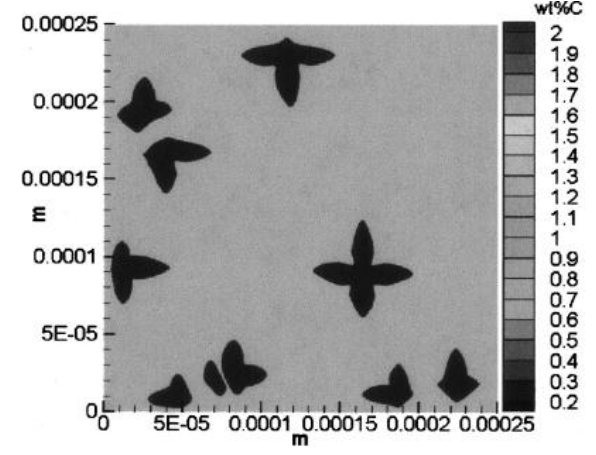


- Dendrite morphology at 0.1 s with different cooling rates caused by different BCs, and comparison to published results
- Higher cooling rate enhances the formation of secondary arm

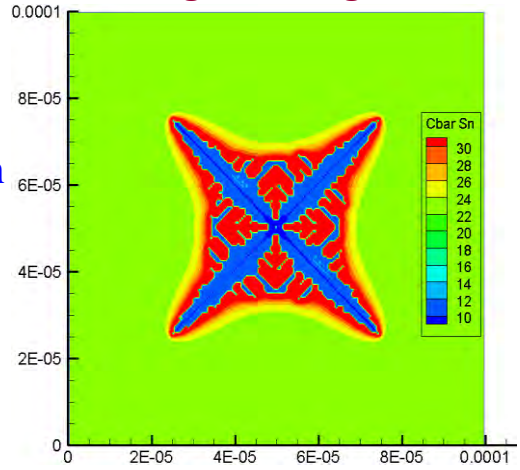
Low cooling rate



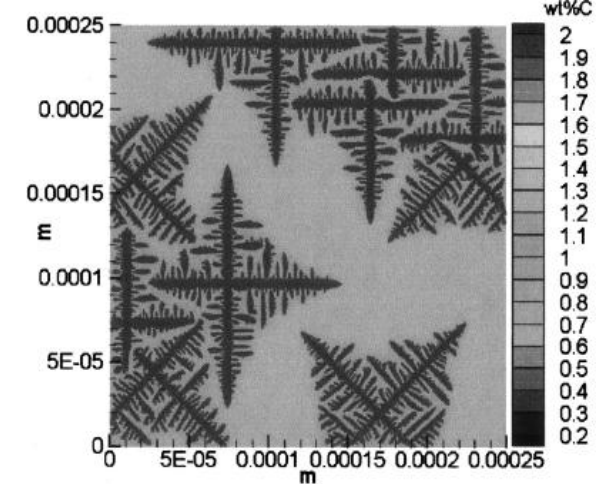
Low cooling rate



High cooling rate



High cooling rate

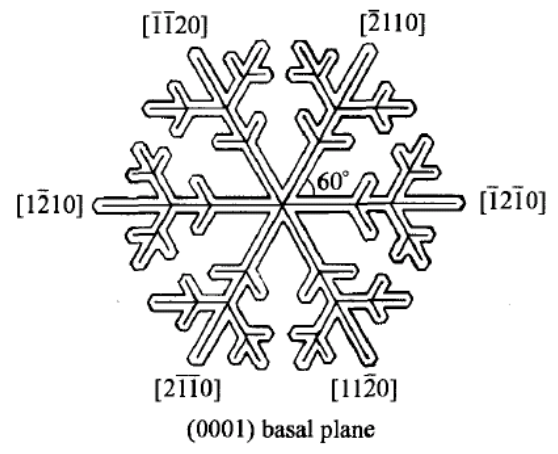


Stefanescu et al, 2003

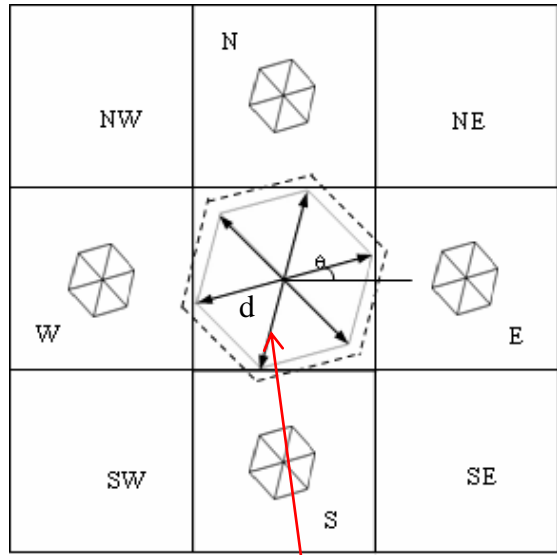
Dendritic Solidification of Cubic and Hexagonal Materials

Cellular automaton

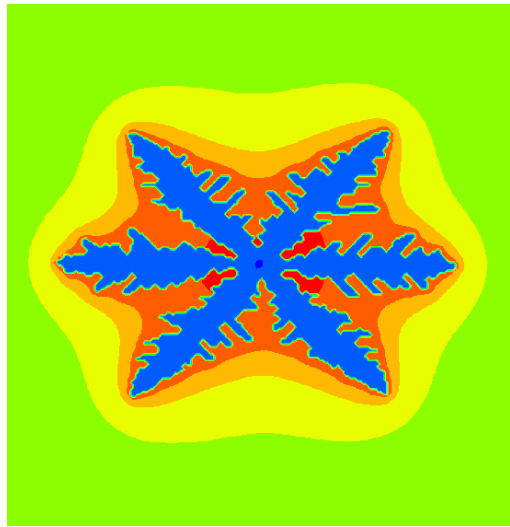
HCP crystal structure with six-fold symmetry



? Problem of result: The angle between the primary arms is not 60 degree. It grows aligning with the axis of the mesh, or at 45 degree



Growth in six directions



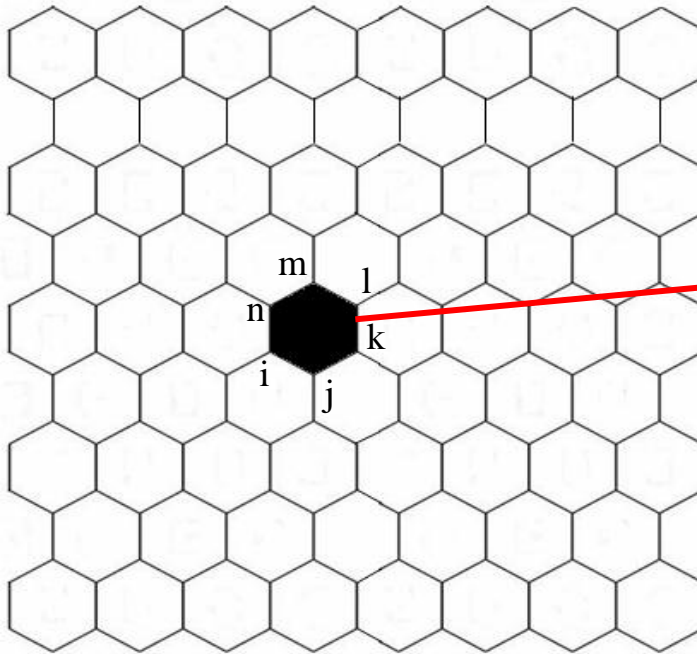
Map of composition

Cellular automaton

- ❑ CA – grid dependent anisotropy, (*Krane , et al. 2008*)
- ❑ Hexagonal mesh reduces grid-induced anisotropy, (*Grest, et al. 1985*)

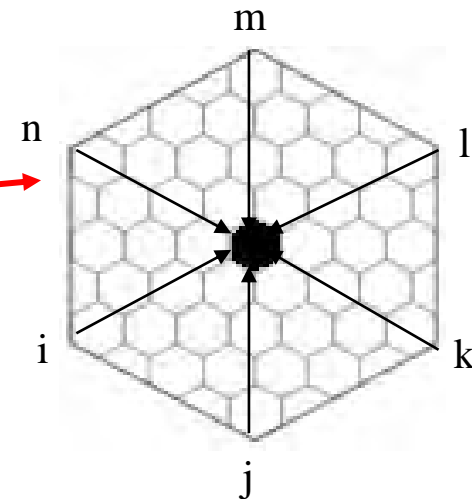
2D hexagonal mesh is generated to simulate the HCP crystal material

FE Mesh



Hexagonal mesh for heat transfer

CA Cell



Mass transfer and dendrite growth

Dendritic Solidification of Cubic and Hexagonal Materials

Phase-field model

Kobayashi's Model – Pure Materials (R. Kobayashi, Physica D 63 (1993) 410-423)

$$F(\phi, m) = \int_V (f_{local}(\phi, m) + f_{grad}(\phi)) dV$$

$$f_{local}(\phi, m) = \int_0^\phi \phi(\phi - 1)(\phi - \frac{1}{2} + m) d\phi$$

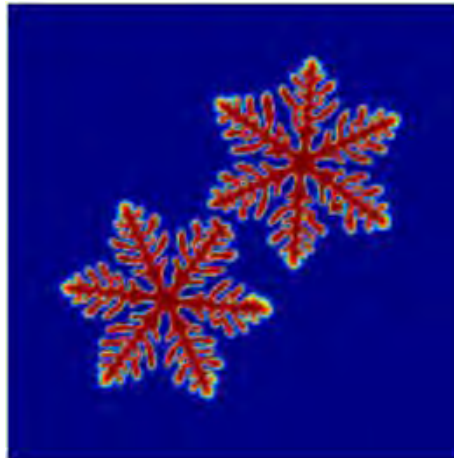
$$f_{grad}(\phi) = \kappa^2 (\nabla \phi)^2 / 2$$

$$\alpha(\theta) = \bar{\alpha} \sigma(\theta)$$

$$m(T) = (\alpha / \pi) \tan^{-1}[\gamma(T_e - T)]$$

$$\sigma(\theta) = 1 + \delta \cos[j(\theta - \theta_0)]$$

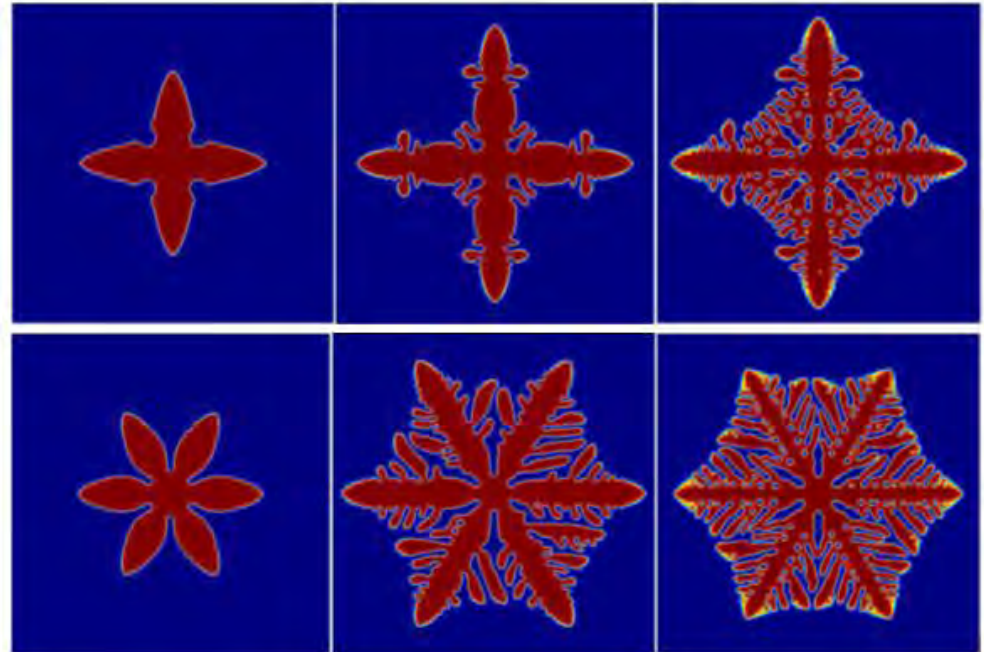
$$\theta = \tan^{-1} \left(\frac{\partial \phi / \partial y}{\partial \phi / \partial x} \right)$$



$$\varepsilon \frac{\partial \phi}{\partial t} = - \frac{\delta F}{\delta \phi} \quad \text{Time-dependent Ginzburg-Landau (TDGL) equation}$$

$$\varepsilon \frac{\partial \phi}{\partial t} = - \frac{\partial}{\partial x} \left(\alpha \alpha' \frac{\partial \phi}{\partial y} \right) + \frac{\partial}{\partial y} \left(\alpha \alpha' \frac{\partial \phi}{\partial x} \right) + \nabla \cdot (\alpha^2 \nabla \phi) + \phi(1 - \phi) \left(\phi - \frac{1}{2} + m \right)$$

$$\frac{\partial T}{\partial t} = \nabla^2 T + K \frac{\partial \phi}{\partial t}$$



Dendritic Solidification of Cubic and Hexagonal Materials

For a binary alloy, the Gibbs-Thomson equation for an isotropic surface energy can be written as

$$\frac{1}{\mu|\nabla\phi|} \frac{\partial\phi}{\partial t} = T_m - T + m_l C_l - \Gamma \nabla \cdot \mathbf{n} \quad \mathbf{n} = -\frac{\nabla\phi}{|\nabla\phi|}$$

Diffusion coef.

$$\tilde{D} = D_s + (D_l - D_s) \frac{1-\phi}{1-\phi+k\phi}$$

Density

$$\tilde{\rho} = \rho_s + (\rho_l - \rho_s) \frac{1-\phi}{1-\phi+k\phi}$$

$$(I) \quad \frac{\partial C}{\partial t} = \nabla \cdot \tilde{D} \left[\nabla C - \frac{(1-k)C}{1-\phi+k\phi} \nabla\phi \right]$$

$$(II) \quad \frac{\partial T}{\partial t} = k \cdot \nabla^2 T + \frac{L}{\tilde{\rho} C_P} \frac{\partial\phi}{\partial t}$$

$$(III) \quad \frac{\partial\phi}{\partial t} = \mu\Gamma \left[-\frac{\partial}{\partial x} \left(\alpha\alpha' \frac{\partial\phi}{\partial y} \right) + \frac{\partial}{\partial y} \left(\alpha\alpha' \frac{\partial\phi}{\partial x} \right) + \nabla \cdot (\alpha^2 \nabla\phi) - \frac{\phi(1-\phi)(1-2\phi)}{\lambda^2} \right] + \mu(T_m - T + m_l C_l) \frac{\phi(1-\phi)}{\lambda}$$

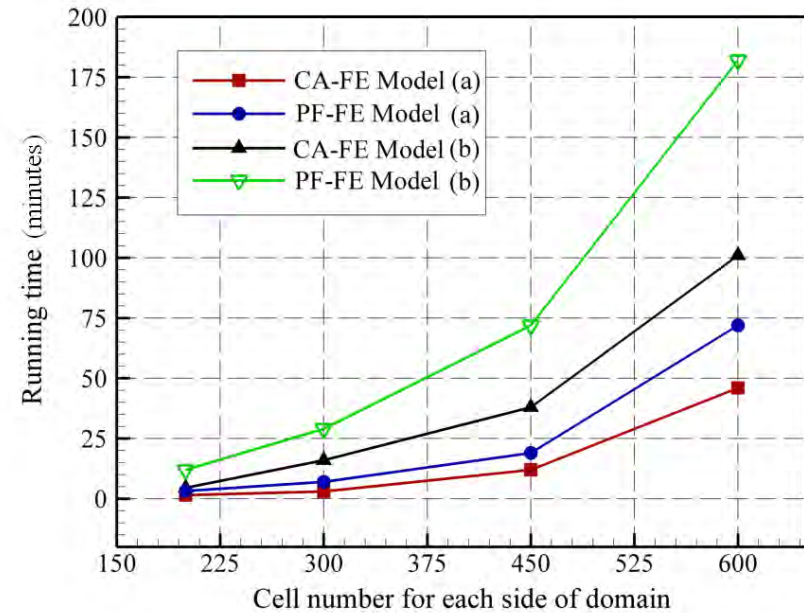
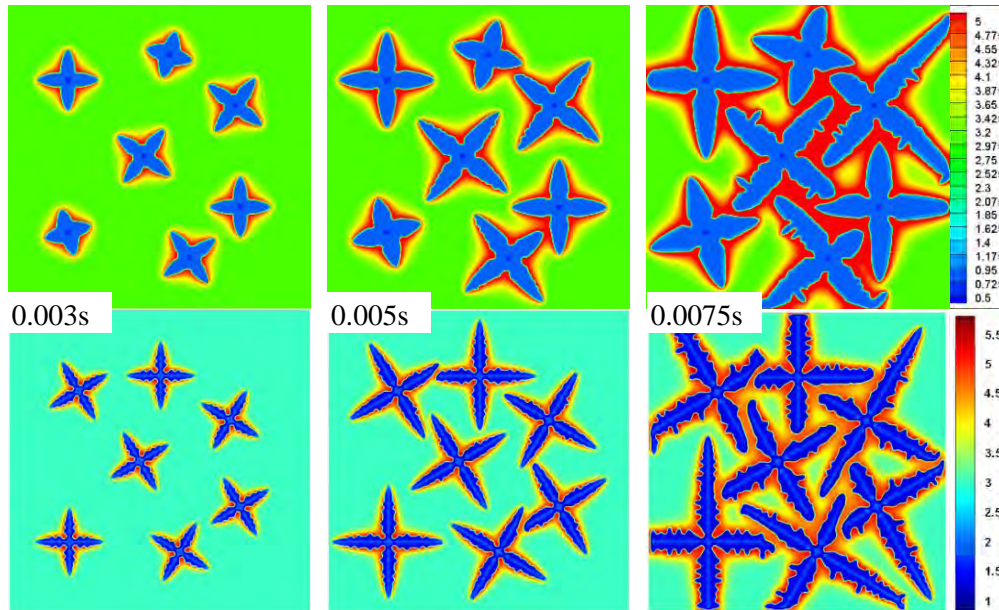
**Governing Equations
Solved by Math Module
COMSOL multiphysics**

Properties of Al-3.0 wt.% Cu alloy

<i>Property</i>	<i>Value</i>
Thermal expansion coefficient (β_T)	$-2.6 \times 10^{-5} \text{ K}^{-1}$
Density	2475 kg m^{-3}
Diffusivity of alloy elements in liquid (D_l)	$3.0 \times 10^{-9} \text{ m}^2 \text{ s}^{-1}$
Diffusivity of alloy elements in solid (D_s)	$3.0 \times 10^{-13} \text{ m}^2 \text{ s}^{-1}$
Thermal conductivity	$30 \text{ J K}^{-1} \text{ m}^{-1} \text{ s}^{-1}$
Average specific heat	$500 \text{ J kg}^{-1} \text{ K}^{-1}$
Latent heat of fusion (L)	$3.76 \times 10^4 \text{ J kg}^{-1}$
Gibbs-Thomson coefficient	$2.4 \times 10^{-7} \text{ K}\cdot\text{m}$
Liquidus slope	-2.6 K/wt pct
Partition ratio	0.17
Melting temperature of pure substance	933.6 K

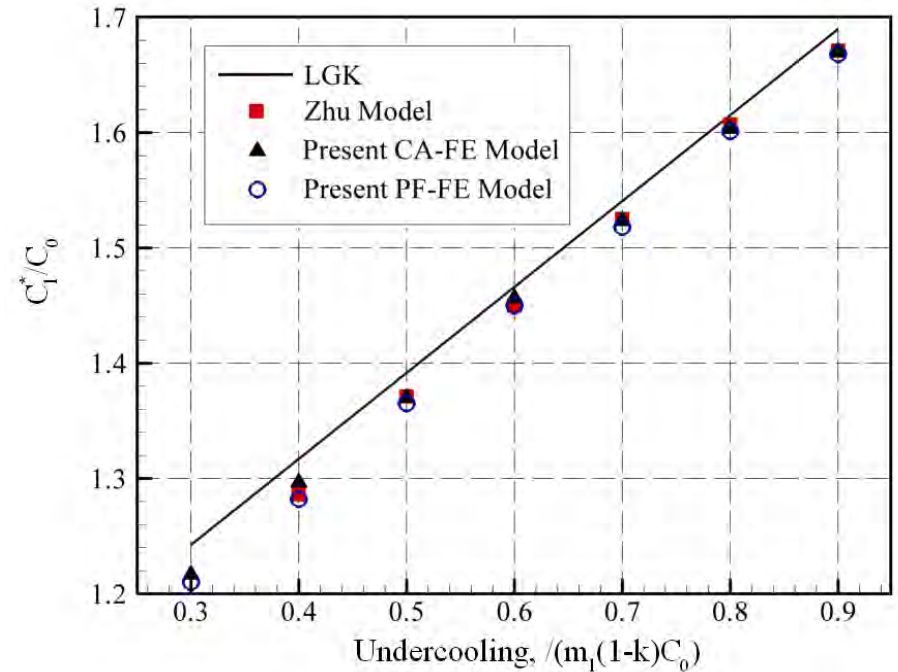
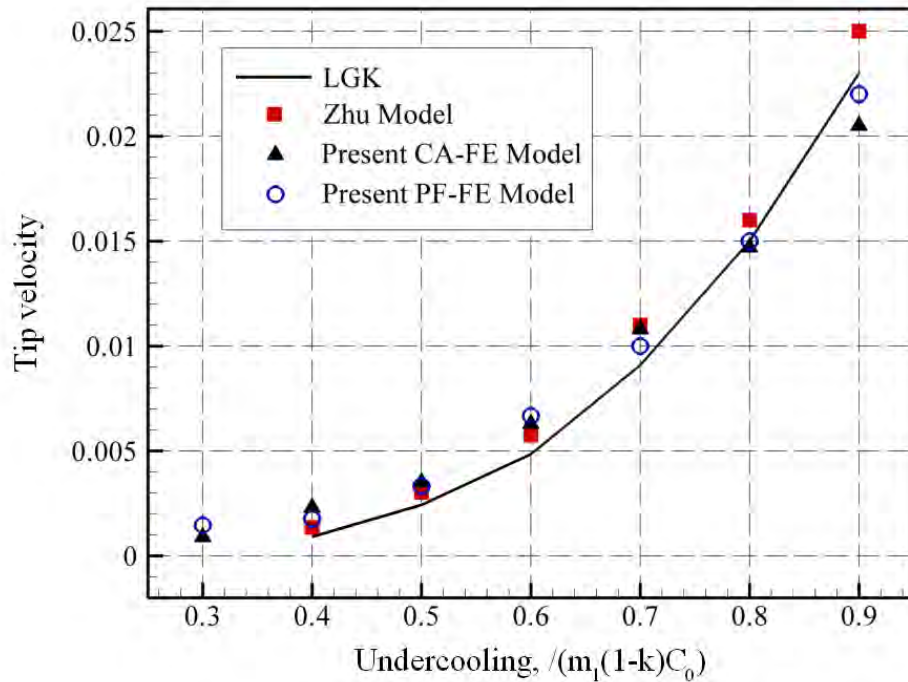
Cellular automaton versus Phase-field model

Al-3.0 wt.% Cu alloy



Cellular automaton versus Phase-field model

Al-3.0 wt.% Cu alloy



MgAZ91 alloy properties

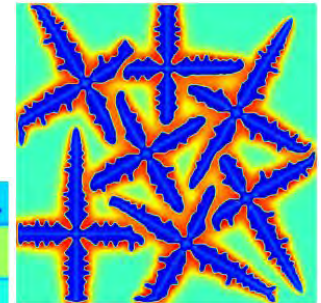
<i>Property</i>	<i>Value</i>
Thermal expansion coefficient (β_T)	$-2.6 \times 10^{-5} \text{ K}^{-1}$
Density of liquid (ρ_l)	1650 kg m^{-3}
Density of solid (ρ_s)	1750 kg m^{-3}
Viscosity (μ)	$2 \times 10^{-3} \text{ N s m}^{-2}$
Diffusivity of alloy elements in liquid (D_l)	$5.0 \times 10^{-9} \text{ m}^2 \text{ s}^{-1}$
Diffusivity of alloy elements in solid (D_s)	$5.0 \times 10^{-13} \text{ m}^2 \text{ s}^{-1}$
Thermal conductivity in liquid (λ_l)	$80 \text{ J K}^{-1} \text{ m}^{-1} \text{ s}^{-1}$
Thermal conductivity in solid (λ_s)	$105 \text{ J K}^{-1} \text{ m}^{-1} \text{ s}^{-1}$
Average specific heat of liquid (c_l)	$1350 \text{ J kg}^{-1} \text{ K}^{-1}$
Average specific heat of solid (c_s)	$1200 \text{ J kg}^{-1} \text{ K}^{-1}$
Latent heat of fusion (L)	$3.7 \times 10^5 \text{ J kg}^{-1}$
Liquidus temperature (T_R)	868 K
Eutectic temperature (T_E)	705 K
Gibbs-Thomson coefficient	$2.0 \times 10^{-7} \text{ K}\cdot\text{m}$



➤ Dendritic Solidification of Cubic and Hexagonal Lightweight Materials

- **Significance:** - multi-component alloys solidification was developed
- multiple-arbitrary orientated hexagonal dendrites

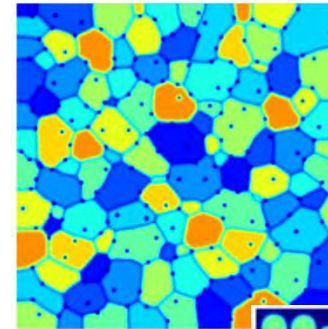
M. Asle Zaeem et al., Mater. Manuf. Processes (2011).



➤ Grain Growth in Polycrystalline Materials

- **Significance:** anisotropic grain boundary energy incorporated in phase-field model

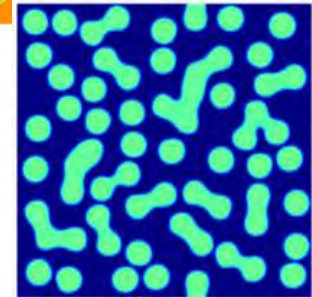
M. Asle Zaeem et al., Comput. Mater. Sci. 50 (8) (2011) 2488-2492.



➤ Phase transformation in binary alloys

- **Significance:** maps of transformations of binary multilayers

M. Asle Zaeem et al., J. Phase Equilib. Diff. 32 (2011) 302-308.

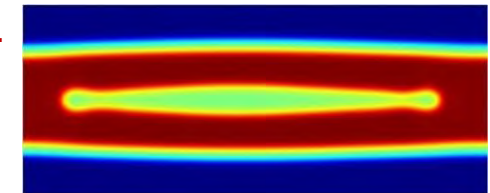


➤ Morphological Instabilities in Multilayers

- **Significance:** maps of transformations of binary multilayers

M. Asle Zaeem & S. Mesarovic, Comput. Mater. Sci. 50 (3) (2011) 1030-1036.

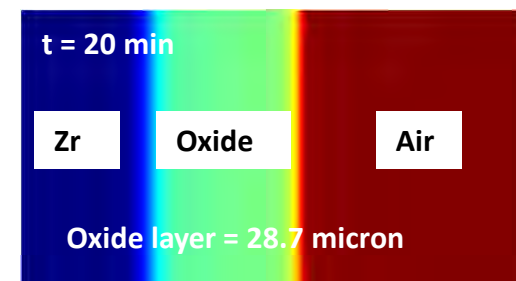
M. Asle Zaeem et al., Modern Physics Letters B 25 (2011) 1591-1601.



➤ Oxidation of Zirconium Alloys in Nuclear Power Plants

- **Significance:** kinetics of oxidation was captured

M. Asle Zaeem et al., J. Nuclear Mater. (2011)-submitted



➤ **3D phase-field finite-element modeling of solidification**



➤ **Study the effects of adding new elements on dendrite shape and spacing (alloys design)**

➤ **Interaction between bifilms and dendrites: oxide bifilms initiate defects after casting**

➤ **Study crystallization of polymers**

Thank you!